

Intermolecular Forces

Intermolecular Forces ie VWF = Attractive Intermolecular forces

Types:

- 1. Dispersion Forces = London Forces**
- 2. Dipole-Dipole Forces**
- 3. Dipole Induced Dipole Forces**
- 4. Hydrogen Bond [strong Dipole Dipole interaction Forces]**

Attractive Forces Between ions and dipoles are known as Ion-Dipole Forces and these are not VWF, Exp: Na^+ and Cl^- H^+