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## Intermolcular Forces

Intermoleular Forces ie VWF = Attractive Intermolecular forces

**Types:** 1. Dispersion Forces = London Forces

- 2. Dipole-Dipole Forces
- 3. Dipole Induced Dipole Forces
- 4. Hydrogen Bond [strong Dipole Dipole interaction Forces]

Atractive Forces Between ions and dipoles are known as Ion-Dipole Forces and these are not VWF, Exp: Na<sup>+</sup> and Cl<sup>-</sup> H<sup>+</sup>